Effect of molecular dissociation on the exchange-correlation Kohn-Sham potential

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The effect of molecular dissociation on the exchange-correlation Kohn-Sham potential \( v_{xc} \) has been studied by the construction of \( v_{xc} \) from the \textit{ab initio} correlated density \( \rho \) for the monohydrides \( XH \) (\( X = \text{Li}, \text{B} \)) at several bond distances \( R(X-H) \). The molecular dissociation manifests itself in the formation of a characteristic peak of \( v_{xc} \) in the bonding region. The partially integrated conditional probability amplitude \( \Phi(x_1,x_2,\ldots,x_N|\vec{r}_1) \) has been used to analyze the behavior of \( v_{xc} \) by means of a partitioning into various components: the potential of the exchange-correlation hole \( v_{xc}^{\text{hole}} \), the kinetic component \( v_{xc,\text{kin}} \) and the "response" component \( v_{\text{resp}} \). These components have been constructed from \textit{ab initio} correlated first- and second-order density matrices. The peak of \( v_{xc} \) in the bonding region has been represented as a combination of the corresponding peak of \( v_{xc,\text{kin}} \) and the positive buildup of \( v_{\text{resp}} \) around the more electronegative atom \( H \). Using the conditional amplitude analysis, asymptotical expressions have been obtained for \( v_{\text{resp}} \) and its positive buildup for the general case of a heteroatomic molecule \( AB \). The dependence of the Kohn-Sham energy characteristics such as the kinetic energy of noninteracting particles \( T_\gamma \), the kinetic part of the exchange-correlation energy \( T_{\text{xc}} \), and the energy of the highest occupied molecular orbital \( e_\text{HOMO} \) on the bond distance has been studied. The results obtained have been compared with those for the homatomic two-electron \( H_2 \) molecule.

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I. INTRODUCTION

Construction of the exchange-correlation Kohn-Sham (KS) potential \( v_{xc} \) and orbitals \( \{\phi_i\} \) from the correlated \textit{ab initio} electron density \( \rho \) of an \( N \)-electron system becomes a promising field within density-functional theory (DFT). Starting with the simple special case of the two-electron atomic systems [1] (and, more recently [2,3]), examples of \( v_{xc} \) were subsequently constructed for the three-electron \( \text{Li} \) atom and light closed-shell atoms \( \text{Be} \) and \( \text{Ne} \) [4–10], and then for all atoms \( \text{He} \) through \( \text{Ar} \) [11,12].

The construction of \( v_{xc} \) was extended to molecules, starting also with the special case of the two-electron \( H_2 \) molecule [2]. Then, \( v_{xc} \) was constructed for \( \text{LiH} \) [13] and for \( \text{BH}, \text{HF} \), and \( \text{N}_2 \) [14,15]. In [2,16] a partitioning scheme was proposed for the analysis of \( v_{xc} \), which is based on the partially integrated conditional probability amplitude \( \Phi(x_1,x_2,\ldots,x_N|\vec{r}_1) \) of the total ground-state wave function \( \Psi(0|x_1,x_2,\ldots,x_N) \). Various components of \( v_{xc} \) are obtained within this scheme from \( \rho \) and also from the correlated first- and second-order density matrices \( \rho(\vec{r}_1,\vec{r}_2) \) and \( \rho_2(\vec{r}_1,\vec{r}_2) \). Using a combination of some components of \( v_{xc} \) a scheme of construction of the exchange-correlation energy density \( e_{xc} \) from \( \rho, \rho(\vec{r}_1,\vec{r}_2) \) and \( \rho_2(\vec{r}_1,\vec{r}_2) \) was proposed in [18] and examples of \( e_{xc} \) were obtained for \( \text{He}, \text{H}_2 \) [2] and also for \( \text{LiH}, \text{BH}, \text{HF} \) [15].

One of the key problems of the molecular KS theory is the effect of various molecular processes on the KS characteristics. The simplest unimolecular process is the dissociation of a diatomic molecule and an interesting topic is the evolution of the KS energy characteristics and the shape of \( v_{xc} \) with increasing bond distance. Hitherto, \( v_{xc} \) has been constructed for the diatomic molecules only at their equilibrium bond lengths \( R_e \). The only exception is the special case of the homatomic two-electron \( H_2 \) molecule, for which the KS solution was obtained in a straightforward way from \( \rho \) at several distances \( R(\text{H–H}) \) [2]. As a matter of fact, even in the latter case the total \( v_{xc} \) has not been presented, only its various components.

In this paper the effect of molecular dissociation on \( v_{xc} \) is investigated by comparison of \( v_{xc} \) and its components constructed from \textit{ab initio} first- and second-order density matrices for the heteroatomic molecules \( \text{LiH} \) and \( \text{BH} \) at several bond distances \( R(X-H) \). Specific dissociation effects for the heteroatomic molecules, in particular, the positive buildup of \( v_{xc} \) around the more electronegative atom are established and interpreted in terms of the partially integrated conditional probability amplitude \( \Phi(x_1,x_2,\ldots,x_N|\vec{r}_1) \). The dependence of the KS energy characteristics such as the kinetic energy of noninteracting particles \( T_\gamma \), the kinetic part of the correlation energy \( T_{\text{xc}} \), and the energy of the highest occupied molecular orbital (HOMO) \( e_\text{HOMO} \) on the bond distance is investigated. The results for the heteroatomic molecules are compared with those for the \( H_2 \) molecule.

II. PARTITIONING OF \( v_{xc} \)

The Kohn-Sham exchange-correlation potential \( v_{xc}([\rho];\vec{r}) \) is a part of the total KS potential \( v_{\gamma}([\rho];\vec{r}) \)

\[
\{ -\frac{1}{2} \nabla^2 + v_{\gamma}([\rho];\vec{r}) \} \phi_{\gamma}(\vec{r}) = e_{\gamma} \phi_{\gamma}(\vec{r}),
\]

\[
\sum_{i=1}^{N} |\phi_{\gamma}(\vec{r})|^2 = \rho(\vec{r}),
\]

or, more precisely, a component of the potential arising from the two-particle electron interaction \( v_{\text{el}}([\rho];\vec{r}) \)

\[
v_{\gamma}(\vec{r}) = v_{\text{ex}}(\vec{r}) + v_{\text{el}}(\vec{r}),
\]
\begin{equation}
    v_{el}(\vec{r}) = v_H(\vec{r}) + v_{xc}(\vec{r}),
\end{equation}

which represents the local effect of electron exchange and Coulomb correlation in the one-electron KS equations (1). Here \( v_{ext} \) is the external potential, \( v_H \) is the Hartree potential of the electrostatic electron repulsion, \( N \) is the number of electrons in the system and the occupied KS orbitals \( \phi_j \) yield the total electron density \( \rho \) via Eq. (2). \( v_{xc}(\rho;\vec{r}) \) is defined in DFT as the functional derivative of the exchange-correlation energy \( E_{xc}[\rho] \) with respect to \( \rho(\vec{r}) \).

\begin{equation}
    v_{xc}(\rho;\vec{r}) = \frac{\partial E_{xc}[\rho]}{\partial \rho(\vec{r})}.
\end{equation}

For the results of this paper it is essential to use the partitioning [2] of \( v_{xc} \) in terms of the conditional probability amplitude \( \Phi(s_1, x_2, \ldots, x_N | \vec{r}_1) \) [17] of the total ground-state wave function \( \Psi_0(x_1, x_2, \ldots, x_N) \) \{\( s_i \) = \{\( r_i, s_i \), \( \vec{r}_i \) \} \} are the spin variables

\begin{equation}
    \Phi(s_1, x_2, \ldots, x_N | \vec{r}_1) = \frac{\Psi_0(x_1, x_2, \ldots, x_N)}{\sqrt{\rho(\vec{r}_1)} \cdot N}.
\end{equation}

\( \Phi(s_1, x_2, \ldots, x_N | \vec{r}_1) \) embodies all effects of electron correlation (exchange as well as Coulomb) in that its square is the probability distribution of the remaining \( N-1 \) electrons associated with positions \( x_2, \ldots, x_N \) when one electron is known to be at \( r_1 \). One can define also the conditional probability amplitude \( \Phi_s(s_1, x_2, \ldots, x_N | \vec{r}_1) \) of the one-determinant wave function \( \Psi_s(x_1, x_2, \ldots, x_N) \) built from the occupied KS orbitals \( \phi_j(\vec{r}_i) \).

\begin{equation}
    \Phi_s(s_1, x_2, \ldots, x_N | \vec{r}_1) = \frac{\Psi_s(x_1, x_2, \ldots, x_N)}{\sqrt{\rho(\vec{r}_1)} \cdot N}.
\end{equation}

To partition \( v_{xc} \) according to [2], one has to start with the stationary \( N \)-electron Schrödinger equations for \( \Psi(x_1, x_2, \ldots, x_N) \) and \( \Psi_s(x_1, x_2, \ldots, x_N) \)

\begin{equation}
    H^N\Psi_0 = E_0^N \Psi_0,
\end{equation}

\begin{equation}
    H^N\Psi_s = E_s^N \Psi_s
\end{equation}

and partition the corresponding Hamiltonians as

\begin{equation}
    H^N = -\frac{1}{2} \nabla_1^2 + v_{ext}(\vec{r}_1) + \sum_{j=2}^N \frac{1}{|\vec{r}_1 - \vec{r}_j|} + H^{N-1},
\end{equation}

\begin{equation}
    H^{N-1} = \sum_{j=2}^N \left\{ -\frac{1}{2} \nabla_j^2 + v_{ext}(\vec{r}_j) + \sum_{k \neq j}^N \frac{1}{|\vec{r}_j - \vec{r}_k|} \right\},
\end{equation}

\begin{equation}
    H_s^N = -\frac{1}{2} \nabla_1^2 + v_{ext}(\vec{r}_1) + v_{el}(\vec{r}_1) + H_s^{N-1},
\end{equation}

\begin{equation}
    H_s^{N-1} = \sum_{j=2}^N \left\{ -\frac{1}{2} \nabla_j^2 + v_{ext}(\vec{r}_j) + v_{el}(\vec{r}_j) \right\}.
\end{equation}

Inserting (6), (7), and (10)–(13) into (8) and (9), we get

\begin{equation}
    \left[ -\frac{1}{2} \nabla_1^2 + v_{ext}(\vec{r}_1) + \sum_{j=2}^N \frac{1}{|\vec{r}_1 - \vec{r}_j|} + H^{N-1} \right] \sqrt{\rho(\vec{r}_1)} / N \Psi_0(s_1, x_2, \ldots, x_N | \vec{r}_1) = E_0^N \sqrt{\rho(\vec{r}_1)} / N \Phi(s_1, x_2, \ldots, x_N | \vec{r}_1),
\end{equation}

\begin{equation}
    \left[ -\frac{1}{2} \nabla_1^2 + v_{ext}(\vec{r}_1) + v_{el}(\vec{r}_1) + H_s^{N-1} \right] \sqrt{\rho(\vec{r}_1)} / N \Phi_s(s_1, x_2, \ldots, x_N | \vec{r}_1) = E_s^N \sqrt{\rho(\vec{r}_1)} / N \Phi_s(s_1, x_2, \ldots, x_N | \vec{r}_1).
\end{equation}

To obtain one-electron equations, both sides of Eqs. (14) and (15) are multiplied by \( \Phi^*(s_1, x_2, \ldots, x_N | \vec{r}_1) \) and \( \Phi_s^*(s_1, x_2, \ldots, x_N | \vec{r}_1) \), respectively, and integrated over the coordinates \( s_1, x_2, \ldots, x_N \). Then, the ground-state energy \( E_0^{N-1} \) of the \((N-1)\) electron system with the same external potential \( v_{ext}(\vec{r}) \) is subtracted from both sides of Eq. (14), while the analogous KS energy \( E_s^{N-1} \)

\begin{equation}
    E_s^{N-1} = \int \Psi_s^{N-1} (\vec{x}_2, \ldots, \vec{x}_N) H_s^{N-1} \Psi_s^{N-1} \times (\vec{x}_2, \ldots, \vec{x}_N) \, d\vec{x}_2 \cdots d\vec{x}_N
\end{equation}

is subtracted from both sides of Eq. (15). Here the \((N-1)\) electron KS determinant \( \Psi_s^{N-1} \) is generated from the \( N \) electron KS determinantal wave function \( \Psi_s \) defined above by the annihilation of one electron from the highest occupied orbital \( \phi_N \). After these operations, making use of the normalization properties of the conditional amplitudes

\begin{equation}
    \int \Phi^*(s_1, x_2, \ldots, x_N | \vec{r}_1) \Phi(s_1, x_2, \ldots, x_N | \vec{r}_1) \, ds_1 \, d\vec{x}_2 \cdots d\vec{x}_N = 1,
\end{equation}

one can obtain from (14) and (15) two equivalent forms of the Euler-Lagrange equation for the square root of the electron density

\begin{equation}
    \left\{ -\frac{1}{2} \nabla^2 + v_{ext}(\vec{r}) + v_H(\vec{r}) + v_{xc}(\vec{r}) + v_{kin}(\vec{r}) + v_s^{N-1}(\vec{r}) \right\} \times \sqrt{\rho(\vec{r})} = \mu \sqrt{\rho(\vec{r})},
\end{equation}

\begin{equation}
    \left\{ -\frac{1}{2} \nabla^2 + v_{ext}(\vec{r}) + v_H(\vec{r}) + v_{xc}(\vec{r}) + v_s^{N-1}(\vec{r}) \right\} \times \sqrt{\rho(\vec{r})} = \mu \sqrt{\rho(\vec{r})}.
\end{equation}
where $\mu = E_0^N - E_0^{N-1}$ is the first vertical ionization energy of the system, which in its turn is equal to the energy of the highest occupied orbital $\mu = \epsilon_s = E_s^N - E_s^{N-1}$.

In Eq. (19) $v_{xc}^{\text{hole}}, v_{\text{kin}}$, and $v_{s}^{N-1}$ are the local potentials obtained from the partially integrated conditional amplitude $\Phi(s_1, x_2, \ldots, x_N|\tilde{r}_1)$, which represent various characteristics of the electron correlation. $v_{xc}^{\text{hole}}$ is the potential of the exchange-correlation hole

$$v_{xc}^{\text{hole}}(\tilde{r}_1) = \int \Phi^*(s_1, x_2, \ldots, x_N | \tilde{r}_1)$$

$$\times \left[ \sum_{j=2}^{N} \frac{1}{|\tilde{r}_1 - \tilde{r}_j|} \right] \Phi(s_1, x_2, \ldots, x_N | \tilde{r}_1)$$

$$\times ds_1 d\tilde{x}_2 \cdots d\tilde{x}_N - v_H(\tilde{r})$$

$$= \int \frac{\rho_s(\tilde{r}_1, \tilde{r}_2) - \rho(\tilde{r}_1) \rho(\tilde{r}_2)}{|\tilde{r}_1 - \tilde{r}_2|} \frac{d\tilde{r}_2}{\rho(\tilde{r}_1)}$$

$$= \int \frac{\rho(\tilde{r}_2) [g(\tilde{r}_1, \tilde{r}_2) - 1]}{|\tilde{r}_1 - \tilde{r}_2|} d\tilde{r}_2, \quad (21)$$

where $\rho_s(\tilde{r}_1, \tilde{r}_2)$ and $g(\tilde{r}_1, \tilde{r}_2)$ are the diagonal part of the second-order density matrix and the pair-correlation function with the electron interaction $\lambda r_{12}$ at full strength $\lambda = 1$. $v_{\text{kin}}$ is the kinetic component

$$v_{\text{kin}}(\tilde{r}_1) = \frac{1}{2} \left\{ \nabla_1 \Phi(s_1, x_2, \ldots, x_N | \tilde{r}_1) \right\}_s^2 ds_1 d\tilde{x}_2 \cdots d\tilde{x}_N$$

$$= \frac{\nabla_1 \rho(\tilde{r}_1, \tilde{r}_1)}{2 \rho(\tilde{r}_1)} - \frac{\nabla_1 \rho(\tilde{r}_1))}{8 \rho^2(\tilde{r}_1)}, \quad (22)$$

where $\rho(\tilde{r}_1, \tilde{r}_1)$ is the first-order density matrix, $v_{\text{kin}}$ reflects the magnitude of change in $\Phi$ with changing $\tilde{r}_1$ (so it is a measure of the change in the correlation hole with variations of the reference position $\tilde{r}_1$). $v_{s}^{N-1}(\tilde{r})$ is the energy expectation value $E_{s}^{N-1}(\tilde{r})$ of the system of $(N-1)$ electrons described by the conditional amplitude

$$E_{s}^{N-1}(\tilde{r}_1) = \int \Phi^*(s_1, x_2, \ldots, x_N | \tilde{r}_1) H_{s}^{N-1} \Phi$$

$$\times (s_1, x_2, \ldots, x_N | \tilde{r}_1) ds_1 d\tilde{x}_2 \cdots d\tilde{x}_N \quad (23)$$

minus the ground-state energy of the $(N-1)$ electron system $E_{s}^{N-1}$

$$v_{s}^{N-1}(\tilde{r}) = E_{s}^{N-1}(\tilde{r}) - E_{s}^{N-1}. \quad (24)$$

The potentials $v_{\text{kin}}$ and $v_{s}^{N-1}$ are obtained, in complete analogy with $v_{\text{kin}}$ and $v_{s}^{N-1}$, from the KS conditional amplitude $\Phi_s(s_1, x_2, \ldots, x_N | \tilde{r}_1)$. Due to the simple one-electron nature of $H_{s}^{N-1}$ and $\Psi_s$, they can be expressed explicitly in terms of the KS orbitals and orbital energies as [16,19]

$$v_{s, \text{kin}}(\tilde{r}_1) = \frac{1}{2} \left\{ \nabla_1 \Phi_s(s_1, x_2, \ldots, x_N | \tilde{r}_1) \right\}_s^2 ds_1 d\tilde{x}_2 \cdots d\tilde{x}_N$$

$$= \frac{1}{2} \sum_{i=1}^{N} \left| \phi_i(\tilde{r}_1) \right|^2 \frac{\phi_i(\tilde{r}_1)}{\rho^{\text{total}}(\tilde{r}_1)}, \quad (25)$$

$$v_{s}^{N-1}(\tilde{r}_1) = \int \Phi^*_s(s_1, x_2, \ldots, x_N | \tilde{r}_1) H_{s}^{N-1} \Phi_s$$

$$\times (s_1, x_2, \ldots, x_N | \tilde{r}_1) ds_1 d\tilde{x}_2 \cdots d\tilde{x}_N - E_{s}^{N-1}$$

$$= \mu - \sum_{i=1}^{N} \epsilon_i \left| \phi_i(\tilde{r}_1) \right|^2 \rho^{\text{total}}(\tilde{r}_1). \quad (26)$$

Equations (19) and (20) provide a partitioning of $v_{xc}$ in terms of the above-mentioned potentials. Equating the left-hand sides of (19) and (20) leads to the following expression for $v_{xc}$

$$v_{xc}(\tilde{r}) = v_{xc}^{\text{hole}}(\tilde{r}) + v_{\text{kin}}(\tilde{r}) + v_{s}^{N-1}(\tilde{r}) - v_{s}^{N-1}(\tilde{r}), \quad (27)$$

where

$$v_{s, \text{kin}}(\tilde{r}) = v_{\text{kin}}(\tilde{r}) - v_{s, \text{kin}}(\tilde{r}). \quad (28)$$

As was shown in [16], the potentials $v_{s}^{N-1}$ and $v_{s}^{N-1}$ can be also expressed in terms of the “response” potentials

$$v_{s}^{N-1}(\tilde{r}) = v_{xc, \text{resp}}^{\text{hole}}(\tilde{r}) + v_{\text{kin}}^{\text{resp}}(\tilde{r}), \quad (29)$$

$$v_{s}^{N-1}(\tilde{r}) = v_{s, \text{kin}}^{\text{resp}}(\tilde{r}). \quad (30)$$

Here the potential $v_{xc, \text{resp}}^{\text{hole}}$ is an integral of the linear “response” $g([\rho]; \tilde{r}_1, \tilde{r}_2), \delta g([\rho]; \tilde{r}_1, \tilde{r}_2)/\delta \rho(\tilde{r}_3)$

$$v_{xc, \text{resp}}^{\text{hole}}([\rho]; \tilde{r}_3) = \frac{1}{2} \int \rho(\tilde{r}_1, \tilde{r}_2) \frac{\delta g([\rho]; \tilde{r}_1, \tilde{r}_2)}{\delta \rho(\tilde{r}_3)} d\tilde{r}_1 d\tilde{r}_2. \quad (31)$$

It is a measure of the sensitivity of the pair-correlation function to density variations. These density variations may be understood in the following way. If the density changes to $\rho + \delta \rho$, then according to the Hohenberg-Kohn theorem this changed density corresponds uniquely to an external potential $v_{\text{ext}} + \delta v_{\text{ext}}$. For the system with external potential $v_{\text{ext}} + \delta v_{\text{ext}}$ we have the corresponding Kohn-Sham system and the pair-correlation function $g([\rho + \delta \rho]; \tilde{r}_1, \tilde{r}_2)$. So the derivative occurring in the response potential (31) may be regarded as the linear response of $g$ to density change $\delta \rho$ caused by potential change $\delta v_{\text{ext}}$. $v_{\text{kin}}^{\text{resp}}$ and $v_{s, \text{kin}}^{\text{resp}}$ are the response of the potentials $v_{\text{kin}}$ and $v_{s, \text{kin}}$ to density variations

$$v_{\text{kin}}^{\text{resp}}([\rho]; \tilde{r}_1) = \int \rho(\tilde{r}_2) \frac{\delta v_{\text{kin}}([\rho]; \tilde{r}_2)}{\delta \rho(\tilde{r}_1)} d\tilde{r}_2, \quad (32)$$

$$v_{s, \text{kin}}^{\text{resp}}([\rho]; \tilde{r}_1) = \int \rho(\tilde{r}_2) \frac{\delta v_{s, \text{kin}}([\rho]; \tilde{r}_2)}{\delta \rho(\tilde{r}_1)} d\tilde{r}_2. \quad (33)$$

The expressions (32) and (33) are obtained from the comparison of Eq. (27) for $v_{xc}$ with its definition (5) as a functional derivative of $E_{xc}([\rho])$. Since the difference between $v_{s}^{N-1}$ and $v_{s}^{N-1}$ accumulates all the response terms, we shall use for it the notation $v_{\text{resp}}$

$$v_{\text{resp}}(\tilde{r}) = v_{s}^{N-1}(\tilde{r}) - v_{s}^{N-1}(\tilde{r}), \quad (34)$$

so that finally we get for $v_{xc}$ the expression
\[ v_{xc}(\vec{r}) = v_{xc}^{\text{hole}}(\vec{r}) + v_{c,\text{kin}}(\vec{r}) + v_{\text{resp}}(\vec{r}). \]  

(35)

It should be noted that the leading term of \( v_{xc} \) is the hole potential \( v_{xc}^{\text{hole}} \). It has been demonstrated that this is, in most regions, the most important part of \( v_{xc} \) indeed [2,18,15]. The fact that the KS potential \( v_s \) incorporates in addition to the potential of the Fermi hole also the potential of the Coulomb hole distinguishes it from the effective potential in the one-electron Hartree-Fock equations which only contains an orbital dependent Fermi-hole contribution. There are many cases where the Hartree-Fock electron density differs strongly from the exact density. Examples are \( \text{H}_2 \) at large distance, and \( \text{N}_2 \) and \( \text{MnO}_4^- \) at equilibrium distance [20]. This is caused primarily by the lack of a Coulomb-hole-potential contribution in the Hartree-Fock potential. In the case of \( \text{MnO}_4^- \) it has been demonstrated [21] that for this reason the Hartree-Fock orbitals are rather distorted, yielding an erroneous heteropolar character to the metal-ligand bond and wrong \( d \)-electron counts when compared to configuration interaction (CI) and complete active space self-consistent field (CASSCF) calculations. The KS orbitals, however, do reflect the homopolar nature of the bonds and lead to \( d \)-electron counts in agreement with the CASSCF calculations. Quite generally it may be argued that, because of the good physics present in the components of the KS potential, the KS orbitals are in no way “inferior” to, e.g., Hartree-Fock orbitals and, in fact, may well be more reliable in qualitative molecular orbital (MO) considerations [22].

In the next section we shall use the presented partitioning scheme and, in particular, formulas for \( v_s^{N-1} \) and \( v_s^{N+1} \) in order to establish and to interpret the features of \( v_{xc} \) which arise in the course of the dissociation of a heteroatomic molecule.

III. ORIGIN OF THE POSITIVE BUILDUP OF \( v_{xc} \) AROUND A MORE ELECTRONEGATIVE ATOM OF A SYSTEM \( AB \)

Dissociation of the heteroatomic bond \( A-B \) produces a spectacular effect on the Kohn-Sham potential \( v_s \), namely, a positive buildup of \( v_s \) around the more electronegative atom \( B \) [23,24]. This can be illustrated with the simple example of two interacting model “one-dimensional hydrogen-like atoms” [24]. A single electron of the model “atom” is bound to the external \( \delta \) function potential \( v_{\text{ext}}(x) = -a \delta(x) \), so that the “atomic” orbital is \( \phi_s(x) = \sqrt{a} \exp(-ax) \) and the ionization energy \( I_A \) is \( a^2/2 \). The single KS orbital of the closed-shell system \( AB \) is constructed as the bonding orbital

\[
\phi(x) = c [\sqrt{a}e^{-a|x|} + \sqrt{b}e^{-b|x-l|}],
\]  

(36)

where \( l \) is the bond length and \( c \) normalizes the total density \( \rho(x) = 2\phi^2(x) \) to two electrons. At large distance \( l \) this construction correctly yields a sum of the “atomic” densities for \( \rho(x) \). The energy \( \epsilon \) of \( \phi(x) \) is equal to minus the total ionization energy of the system, which at large \( l \) values is equal to that of the less electronegative atom \( A \), \( \epsilon = -a^2/2 \).

Figure 1 represents the KS potential \( v_s(x) \) of the system \( AB \) obtained for \( l = 3 \), \( l = 7 \) a.u. and \( x \neq x_A \), \( x \neq x_B \) with the insertion of (36) and \( -I_A = a^2/2 \) into the one-electron KS equation

\[
\Delta v_s = I_B - I_A
\]  

(38)

in the general case of the real three-dimensional heteroatomic system \( AB \) at large bond distances. In [24] the principle of equalization of the KS orbital energies for the fragments \( A \) and \( B \) was used and discontinuities of the derivative of the total energy \( E \) as a function of the particle number \((N+\omega)\) were explored. In [23] arguments concerning the asymptotic behavior in different regions of the highest occupied MO (HOMO) were used to show (38). In [23] the positive buildup was attributed to the exchange-correlation part \( v_{xc} \) of \( v_s \).
In this paper we establish the existence of the positive build-up of \(v_{xc}\) around the more electronegative atom \(B\) of a system \(AB\) directly from the total many-electron wave function \(\Psi(\vec{x}_1, \ldots, \vec{x}_N)\) of the system. Based on the partitioning of \(v_{xc}\) in Sec. II we shall demonstrate that this buildup originates in the response part \(v_{\text{resp}}\) of \(v_{xc}\) and we shall derive an expression for it, making use of the conditional probability amplitude \(\Phi(\vec{x}_1, \ldots, \vec{x}_N|\vec{r}_1)\) [Eq. (6)].

We start with the definition (34) of \(v_{\text{resp}}(\vec{r}_1)\) and we shall analyze the form of \(v_{\text{resp}}(\vec{r}_1)\) in the region of the highest occupied MO (HOMO) \(\phi_N(\vec{r}_1)\). It follows from (26) that \(v_{N}^{N-1}(\vec{r}_1)\) in this region vanishes, since \(\phi_N(\vec{r}_1)\) constitutes the dominant contribution to \(\rho(\vec{r}_1)\), so that \(v_{\text{resp}}(\vec{r}_1)\) reduces effectively to \(v^{N-1}(\vec{r}_1)\)

\[
v_{\text{resp}}(\vec{r}_1) = v^{N-1}(\vec{r}_1) = E^{N-1}(\vec{r}_1) - E_0^{N-1},
\]

which, according to (23), (24), is determined by the function \(\Phi(\vec{x}_1, \ldots, \vec{x}_N|\vec{r}_1)\).

Suppose that the reference electron is in the region \(\Omega_\Lambda\) of the more electronegative atom \(B\), \(\vec{r}_1 \in \Omega_\Lambda\). In this case the conditional amplitude \(\Phi(\vec{x}_1, \ldots, \vec{x}_N|\vec{r}_1)\) describes the \((N-1)\)-electron system \(A^-B^+\) consisting of the neutral atom \(A\) interacting with a cation \(B^+\). This cation will not in general be in the ground state of the \(B^+\) system, but if \(\vec{r}_1\) is actually at significant distance from the electronic cloud of \(B^+\), although still by assumption much closer to \(B\) than to \(A\), it has been established by Katriel and Davidson [26] that \(B^+\) is in that case in its ground state. So in that case at large bond distances \(R(A-B)\) the energy of this system reduces to

\[
E(A-B^+) \approx E_0(A) + E_0(B) + I_B + E_{\text{in}}(A-B^+),
\]

where \(E_0(A)\) and \(E_0(B)\) are the ground-state energies of the atoms \(A\) and \(B\), respectively, \(I_B\) is the ionization energy of the atom \(B\), and \(E_{\text{in}}(A-B^+)\) is the energy of interaction of the atom \(A\) with the cation \(B^+\). If we allow \(\vec{r}_1\) to be in the neighborhood of the other electrons of \(B^+\), it is necessary to take into account that the conditional amplitude will not describe the ground state of \(B^+\). The fact that the system described by \(\Phi\) is “distorted” will correspond to an energy rise \(\Delta E\) with respect to the ground-state energy. We may therefore write the energy \(E^{N-1}(\vec{r}_1 \in \Omega_\Lambda)\) in general as

\[
E^{N-1}(\vec{r}_1 \in \Omega_\Lambda) \approx E_0(A) + E_0(B) + I_B + E_{\text{in}}(A-B^+) + \Delta E(A-B^+;\vec{r}_1).
\]

If \(R(A-B)\) is large and \(\vec{r}_1\) is in the region of the HOMO (i.e., not in the valence-core region of \(B\)), the effect of the electron redistribution incorporated in the last term is expected to be small.

Contrary to this, the ground state of the cation \((AB)^+\) for large \(R(A-B)\) corresponds to the system \(A^-B^+\) of the neutral atom \(B\) interacting with the cation \(A^+\), so that \(E_0^{N-1}\) is expressed as

\[
E_0^{N-1} \approx E_0(A) + E_0(B) + I_A + E_{\text{in}}(A^+-B). \tag{42}
\]

Inserting (41) and (42) into (39), we obtain the following expression for \(v_{\text{resp}}(\vec{r}_1)\):

\[
v_{\text{resp}}(\vec{r}_1 \in \Omega_\Lambda) = [I_B - I_A] + [E_{\text{in}}(A-B^+) - E_{\text{in}}(A^+-B)] + \Delta E(A-B^+;\vec{r}_1). \tag{43}
\]

Suppose now that the reference electron is in the region \(\Omega_\Lambda\) of the less electronegative atom \(A\), \(\vec{r}_1 \in \Omega_\Lambda\). In this case the conditional amplitude describes the \((N-1)\)-electron system \(A^+-B\), disturbed around the reference electron position. If \(R(A-B)\) is large and \(\vec{r}_1\) is in the region of the HOMO, this system is close to the ground state of the cation \((AB)^+\). As a result, only the corresponding correction term \(\Delta E(A-B;\vec{r}_1)\) contributes to \(v_{\text{resp}}(\vec{r}_1)\) in this region

\[
v_{\text{resp}}(\vec{r}_1 \in \Omega_\Lambda) = \Delta E(A^+-B;\vec{r}_1). \tag{44}
\]

From (43) and (44) we can estimate the build-up \(\Delta v_{\text{resp}}\) around the more electronegative atom \(B\)

\[
\Delta v_{\text{resp}} = v_{\text{resp}}(\vec{r}_1 \in \Omega_\Lambda) - v_{\text{resp}}(\vec{r}_1 \in \Omega_A) = [I_B - I_A] + [E_{\text{in}}(A-B^+) - E_{\text{in}}(A^+-B)] + \Delta E(A-B^+;\vec{r}_1) - \Delta E(A^+-B;\vec{r}_1 \in \Omega_A).
\]

The leading term of (45) at large bond distances \(R(A-B)\) is just the difference of the ionization energies of atoms \(A\) and \(B\). Formula (45) demonstrates that the positive build-up \(\Delta v_{xc}=(I_B-I_A)\) emerges in the response part \(v_{\text{resp}}\) of \(v_{xc}\) or, more precisely, in the \(v^{N-1}\) component of \(v_{\text{resp}}\). It originates from the difference between the conditional amplitude distribution \(\Phi(\vec{x}_1, \ldots, \vec{x}_N|\vec{r}_1)\) of \((N-1)\) electrons and the ground-state electron distribution of the cation \((AB)^+\). When \(\vec{r}_1 \in \Omega_\Lambda\), the conditional amplitude distribution corresponds to the system \(A^+-B^+\), while the ground state is the cation \(A^-B\). Thus, the conditional amplitude, embodying the electron correlation which causes the complete exchange-correlation hole to be located around the reference position, leads to a “repulsive” effect on \(v_{xc}\) in \(\Omega_\Lambda\). The KS potential at a point \(\vec{r}_1\) in the energetically favorable region, is shifted upwards by a potential barrier of height \((I_B-I_A)\), which emerges from \(v^{N-1}\), to prevent a too strong localization of electrons in that region.

The terms in the second and third square brackets of Eq. (45) provide corrections to the leading term at large \(R(A-B)\). The second term represents a correction from the atom-atom interaction, which is different for the pairs \(A^+-B\) and \(A^-B^+\). The third term represents a difference between the energy effects of the redistribution of \((N-1)\) electrons of \(A^+-B^+\) and \(A^+-B\) due to the presence of the reference electron position in the outer region of the corresponding charged atom. In other words, the first term brings the main contribution to \(\Delta v_{\text{resp}}\) due to the different ionization of \(A\) and \(B\), the second one brings a correction due to the different polarization of \(A\) and \(B\) by a positive charge, and the third one brings a correction due to the different distortions of the cations \(A^-B^+\) and \(A^+-B\) due to the different “response” to the proximity of the reference electron position. Since this “response” has some relation to the polarizability, and since the polarizability of the less electronegative...
atom A is, in general, higher than that of B, both corrections are expected to have a negative sign, opposite to that of the leading term \([I_B - I_A]\).

All the above-mentioned differences tend to decrease with decreasing electronegativity difference of atoms A and B, and turn into zero for the homatomic molecule \(A_2\) (45), as they should. For \(A_2\) the expression

\[
v_{\text{rexp}}(r_1) = \Delta E(A^+ - A; r_1),
\]

(46)

which is an analogue of (44), is valid for the HOMO region and \(v_{\text{rexp}}(r_1)\) is expected to be small and to have a flat form in this region. This is true, in particular, for \(v_{\text{rexp}}(r_1)\) of the molecule \(H_2\) (see [2] and also the next section).

To sum up, using the partially integrated conditional probability amplitude of the heteroatomic molecule \(AB\), it has been shown that the positive buildup of \(v_{\text{xc}}\) around the electronegative atom B originates in the \(p^{N-1}\) component of the response part of \(v_{\text{xc}}\). From the asymptotical formulas for \(v^{N-1}(r_1)\) in different regions an expression for this buildup has been obtained which, in addition to the leading term \((I_B - I_A)\) contains also the polarization and correlation corrections. Examples of the positive buildup of \(v_{\text{xc}}\) constructed from \textit{ab initio} wave functions for the monohydrides LiH and BH will be presented in the next sections.

IV. CONSTRUCTION OF \(v_{\text{xc}}\)

The scheme of \(v_{\text{xc}}\) construction used in this paper has been already presented and discussed in [13,18] and here we only mention its main points. The correlated wave functions have been obtained with singly and doubly excited configuration interaction (SDCI) calculations of the ground states at the bond distances \(R_e = 1.401\) a.u., \(R = 3.0\) a.u., and \(R = 5.0\) a.u. for \(H_2; R_e = 3.015\) a.u., \(R = 5.0\) a.u., and \(R = 7.0\) a.u. for LiH; \(R_e = 2.330\) a.u., \(R = 4.0\) a.u., and \(R = 5.0\) a.u. for BH. Calculations have been performed within the direct CI approach by means of the ATMOL package [27].

A basis of contracted Gaussian functions [28] has been used with five \(s\)- and two \(p\)-type functions for H, seven \(s\)-, four \(p\)-type functions for Li, seven \(s\)- four \(p\)- and two \(d\)-type functions for B. For H an extra valence polarization \(d\) function with the exponent \(\alpha = 1.0\) and for Li two such functions with the exponents \(\alpha = 0.36\) and \(\alpha = 0.15\) have been used. This basis has been already used for the construction of \(v_{\text{xc}}\) for LiH at the equilibrium bond length [13]. In this paper, in order to better take into account the correlation effects for core electrons, this basis has been augmented for Li and B with two localized polarization \(p\) and two \(d\) functions of the core size, whose exponents were set equal to those of the second most localized contracted \(s\) function of the basis [28].

When compared with the accurate empirical estimates of the Coulomb correlation energies \(E^c_{\text{xc}}\) at the equilibrium distances [29], the decrease in the correlation energy due to inclusion of core polarization functions amounts to 6\% of \(E^c_{\text{xc}}\) for LiH and 14\% for BH, so that the SDCI calculations recover 92\% of \(E^c_{\text{xc}}\) for LiH and 90\% for BH. The potential \(v_{\text{rexp}}(r_1)\) for a given grid \(\{\hat{r}_i\}\) has been calculated by integration (21) of the diagonal part \(\rho_2(\hat{r}_i, \hat{r}_j)\) of the second-order density matrix, the latter has been calculated from the SDCI wave function by means of a special density-functional extension [2,30] of the \textit{ab initio} ATMOL package.

In the case of the \(H_2\) molecule there is only one occupied KS orbital \(\phi_1\), so that an accurate \(v_{\text{xc}}\) is obtained directly from (1) by the replacement of \(\phi_1\) by \(\rho(\mu)^{1/2}\) [2]. For \(XH\) \(v_{\text{xc}}(\hat{r})\) and a set of KS orbitals \(\{\phi_i(\hat{r})\}\) are obtained from the correlated \(\rho(\hat{r})\) with an iterative procedure [10], starting from some initial guess \(v_{\text{el}}^0(\hat{r})\) for \(v_{\text{el}}\)

\[
v_{\text{el}}^0(\hat{r}) = v_{\text{H}}(\hat{r}) + v_{\text{xc}}^0(\hat{r}),
\]

(47)

where \(v_{\text{el}}\) is the Hartree potential and \(v_{\text{xc}}^0\) is an approximate exchange-correlation potential of the form

\[
v_{\text{xc}}^0(\rho; \hat{r}) = v_{XX}(\rho; \hat{r}) + 2 \epsilon^0_{\text{xc}}(\rho; \hat{r}) + 2 \epsilon^\text{xWVN}_{\text{xc}}(\rho; \hat{r}),
\]

(48)

both potentials being calculated with the correlated density \(\rho\). In (48) \(v_{XX}(\rho; \hat{r})\) is the exchange-correlation \(X\alpha\) potential [31], \(\epsilon^0_{\text{xc}}(\rho; \hat{r})\) is the exchange energy gradient correction of Becke [32], and \(\epsilon^\text{xWVN}_{\text{xc}}\) is the local-density approximation (LDA) of Vosko, Wilk, and Nusair [33] for the correlation energy density. The potential (47) has a proper long-range Coulombic asymptotics \(v_{\text{el}}^0(\rho; \hat{r}) \rightarrow -(N - 1)/|\hat{r}|\). For the equilibrium bond length the parameter \(\alpha_{\text{eq}}\) of \(v_{XX}\) is chosen from the following fitting function:

\[
\left[-\frac{1}{2} \nabla^2 + v_{\text{el}}(\hat{r}) + v_{\text{el}}^0(\hat{r})\right] \phi_N(\hat{r}) = -I_N \phi_N(\hat{r}),
\]

(49)

where \(\phi_N\) is the HOMO and \(I_N\) is the experimental ionization energy of the molecule. For larger distances \(R(X-H)\) the parameter \(\alpha\) is varied starting from \(\alpha_{\text{eq}}\) and, finally, the value \(\alpha\) is used, which provides the quickest convergence of the iterative procedure.

At \(m\)-th iteration KS equations (1) are solved with the potential \(v_{\text{el}}^m(\hat{r})\)

\[
v_{\text{el}}^m(\hat{r}) = f_m(\hat{r}) v_{\text{el}}^{m-1}(\hat{r})
\]

(50)

calculated from \(v_{\text{el}}^{m-1}\) of the previous iteration with the correction factor \(f_m\), the latter being defined with the density \(\rho^{m-1}\) from the \((m-1)\)-th iteration and the \textit{ab initio} target density \(\rho\)

\[
f_m(\hat{r}) = \frac{\rho^{m-1}(\hat{r}) + a}{\rho(\hat{r}) + a}
\]

(51)

with the parameter \(a = 0.5\), which smooths out the effect of the remote exponential density tails on the procedure. Then, \(\rho^{m-1}\) in (51) is replaced with \(\rho^m\) obtained at \(m\)-th iteration and this procedure continues unless further iterations cease lowering the difference \(\rho^m(\rho^{m-1}) - \rho(\hat{r})\) in the region of non-vanishing densities. Finally, \(v_{\text{xc}}(\hat{r})\) is obtained by subtracting \(v_{\text{H}}(\hat{r})\) from the resulting potential (50). Construction of \(v_{\text{xc}}\) has been performed in the same basis of MO’s as the SDCI calculations by means of the above-mentioned density-functional extension [2,30] of the ATMOL package. Matrix elements of \(v_{\text{el}}\) in this basis have been calculated using a numerical integration with grids according to Ref. [34].

After 45–50 iterations the procedure has reached its saturation state and further several hundred iterations make changes only within 0.001–0.003 a.u. for the obtained values of \(\epsilon_{\text{xc}}\) and the KS kinetic energy \(T\) and produce basically the same KS orbitals \(\{\phi_i(\hat{r})\}\). However, after 200–250 iterations.
TABLE I. Kohn–Sham energy characteristics (a.u.) for LiH at various bond distances $R$(Li—H) (a.u.).

<table>
<thead>
<tr>
<th>$R$(Li—H)</th>
<th>3.015</th>
<th>5.0</th>
<th>7.0</th>
</tr>
</thead>
<tbody>
<tr>
<td>$\epsilon_{\text{HOMO}}$</td>
<td>$-0.284$</td>
<td>$-0.238$</td>
<td>$-0.206$</td>
</tr>
<tr>
<td>$-I_p$</td>
<td>$-0.283$</td>
<td>$-0.198$</td>
<td></td>
</tr>
<tr>
<td>$T_x$</td>
<td>8.001</td>
<td>7.823</td>
<td>7.826</td>
</tr>
<tr>
<td>$T_{\text{HF}}$</td>
<td>7.993</td>
<td>7.787</td>
<td>7.747</td>
</tr>
<tr>
<td>$T_{\text{CI}}$</td>
<td>8.058</td>
<td>7.880</td>
<td>7.876</td>
</tr>
<tr>
<td>$T_e$</td>
<td>0.057</td>
<td>0.057</td>
<td>0.050</td>
</tr>
<tr>
<td>$T_{e,\text{HF}}$</td>
<td>0.065</td>
<td>0.093</td>
<td>0.129</td>
</tr>
</tbody>
</table>

TABLE II. Kohn-Sham energy characteristics (a.u.) for BH at various bond distances $R$(B—H) (a.u.).

<table>
<thead>
<tr>
<th>$R$(B—H)</th>
<th>2.33</th>
<th>4.0</th>
<th>5.0</th>
</tr>
</thead>
<tbody>
<tr>
<td>$\epsilon_{\text{HOMO}}$</td>
<td>$-0.359$</td>
<td>$-0.346$</td>
<td>$-0.334$</td>
</tr>
<tr>
<td>$-I_p$</td>
<td>$-0.359$</td>
<td>$-0.305$</td>
<td></td>
</tr>
<tr>
<td>$T_x$</td>
<td>25.153</td>
<td>24.883</td>
<td>24.893</td>
</tr>
<tr>
<td>$T_{\text{HF}}$</td>
<td>25.119</td>
<td>24.793</td>
<td>24.757</td>
</tr>
<tr>
<td>$T_{\text{CI}}$</td>
<td>25.252</td>
<td>24.980</td>
<td>24.985</td>
</tr>
<tr>
<td>$T_e$</td>
<td>0.099</td>
<td>0.097</td>
<td>0.092</td>
</tr>
<tr>
<td>$T_{e,\text{HF}}$</td>
<td>0.133</td>
<td>0.187</td>
<td>0.228</td>
</tr>
</tbody>
</table>

(assuming on the type of system and the bond distance) visible and (presumably) artificial long-range oscillations are developed in $v_{xc}$, which do not alter the above-mentioned KS characteristics but disturb the form of $v_{xc}$. For larger bond distances these oscillations arise at earlier iterations than for $R_e$. The development of the (spurious) long-range oscillations of $v_{xc}$ after several hundred iterations seems to be an artifact of the finite basis set used in the molecular calculations, since no such effect appears in the numerical atomic calculations with the procedure (47), (50), (51) [10]. The effect of the finite basis restriction on the constructed $v_{xc}$ has been mentioned in [35]. The possibility of these oscillations follows from the fact that, in principle, an oscillating function can be added to $v_{xc}$, such that its addition does not change matrix elements of $v_{xc}$ in a given finite basis. Though not disturbing the integral characteristics, the development of these oscillations can be recognized by the increasing of the maximal local relative difference

$$\Delta \rho_{\text{max}}^m = \max_{r \in \Omega} \frac{|\rho^m(r) - \rho(r)|}{\rho(r)}$$

(52)

(Ω is the region of nonvanishing densities) between the target density $\rho$ and the density $\rho^m$ starting from a certain $(m+1)$th iteration. In our present calculations, in order to prevent the development of the spurious oscillations, we terminate the iterative procedure at the $m$th iteration, at which the difference (52) attains its minimum for $\rho > 0.1$ a.u. The corresponding $\Delta \rho_{\text{max}}^m$ values are within $0.1$ to $0.2\%$.

The potential $v_{c,\text{kin}}(\vec{r})$ has been calculated from the SDCI first-order density matrix $\rho(\vec{r}', \vec{r})$ and the KS density matrix $\rho(\vec{r}', \vec{r}) = \sum_{n=1}^{N} \phi_n^*(\vec{r}') \phi_n(\vec{r})$ obtained with the iterative procedure discussed above. Finally, $v_{\text{resp}}(\vec{r})$ has been obtained by subtraction of $v_{c,\text{kin}}(\vec{r})$ and $v_{xc}(\vec{r})$ from the constructed $v_{xc}(\vec{r})$. The results of the scheme presented in this section will be discussed in the next sections.

V. Kohn-Sham Energy Characteristics

Tables I and II present various energy characteristics for the KS orbitals $\{\phi_n(\vec{r})\}$ of the LiH and BH for three different bond distances $R$(X—H). The first is the HOMO energy $\epsilon_{\text{HOMO}}$. According to [36–39] an accurate HOMO energy $\epsilon_{\text{HOMO}}$ is equal to minus the ionization potential of the system $I_p$. In the dissociation limit the ionization potential of XH approaches that of the less electronegative atom X, so that $\epsilon_{\text{HOMO}}$ should follow the same trend. One can see from Tables I and II that

\[\epsilon_{\text{HOMO}} \text{ values obtained for XH at the equilibrium distance } R_e(\text{X—H}) \text{ are indeed very close to the experimental } -I_p \text{ values (the latter quantities are placed in Tables I and II just below the former ones). For both systems } \epsilon_{\text{HOMO}} \text{ decreases (in absolute magnitude) with increasing } R \text{ and for the largest } R \text{ value it is not far from } -I_p \text{ value of the X atom (the latter quantities are also placed just below the former ones).}

Tables I and II also present the kinetic energy of the KS system $T_s$

$$T_s = -\frac{1}{2} \sum_{i=1}^{N} \int \phi_i^*(\vec{r}) \nabla^2 \phi_i(\vec{r}) d\vec{r}$$

and the kinetic part of the exchange-correlation energy $T_e$

$$T_e = T_{\text{CI}} - T_s = \int \rho(\vec{r}) v_{c,\text{kin}}(\vec{r}) d\vec{r}.$$  

(54)

$T_s$ is compared with the CI $T_{\text{CI}}$ and the Hartree-Fock $T_{\text{HF}}$ kinetic energies, while $T_e$ is compared with the kinetic part $T_{e,\text{HF}}$ of the conventional correlation energy

$$T_{e,\text{HF}} = T_{\text{CI}} - T_{\text{HF}}.$$  

(55)

In Table III the same kinetic characteristics are presented for the H$_2$ molecule.

In all cases the $T_s$ value is placed in between the corresponding $T_{\text{HF}}$ and $T_{\text{CI}}$ ones

$$T_{\text{HF}} < T_s < T_{\text{CI}}.$$  

(56)

The right-hand inequality of (56) follows from the fact that both $T_s$ and $T_{\text{CI}}$ correspond to the same correlated density $\rho$ and $T_s$, by its definition [40], delivers the minimal kinetic energy for this density. The left-hand inequality reflects the difference between the Hartree-Fock and correlated densities. It is well known that the correlated density is more

TABLE III. Kinetic energy characteristics (a.u.) for H$_2$ at various bond distances $R$(H—H) (a.u.).

<table>
<thead>
<tr>
<th>$R$(H—H)</th>
<th>1.401</th>
<th>3.0</th>
<th>5.0</th>
</tr>
</thead>
<tbody>
<tr>
<td>$T_s$</td>
<td>1.140</td>
<td>0.831</td>
<td>0.955</td>
</tr>
<tr>
<td>$T_{\text{HF}}$</td>
<td>1.125</td>
<td>0.713</td>
<td>0.650</td>
</tr>
<tr>
<td>$T_{\text{CI}}$</td>
<td>1.172</td>
<td>0.872</td>
<td>0.977</td>
</tr>
<tr>
<td>$T_e$</td>
<td>0.032</td>
<td>0.041</td>
<td>0.022</td>
</tr>
<tr>
<td>$T_{e,\text{HF}}$</td>
<td>0.047</td>
<td>0.159</td>
<td>0.327</td>
</tr>
</tbody>
</table>
contracted around the nuclei than the Hartree-Fock one, which is extremely so in poor Hartree-Fock cases like dissociating H₂ [2,20]. Due to this contraction effect of correlation, the minimal energy $T_c$ is still higher than $T_{\text{HFF}}$.

For H₂ and BH $T^{\text{CI}}$ value at the intermediate bond distance is lower than those at the larger and equilibrium distances, while for LiH $T^{\text{CI}}$ values at $R = 5.0$ and 7.0 a.u. are close to each other, both being appreciably lower than that at $R_c = 3.015$ a.u. It is an anticipated trend since, according to the virial theorem formula [41]

$$T = -E_0^N - R \frac{dE_0^N}{dR}, \quad (57)$$

the exact kinetic energy $T$ has a negative contribution from the Hellmann-Feynman forces at lower distances, which vanishes at both $R = R_c$ and $R \to \infty$, so that $T$ as a function of $R$ passes through a minimum at $R > R_c$. For the restricted CI the virial theorem holds only approximately, still $T^{\text{CI}}$ exhibits a similar behavior. The virial theorem is accomplished through a lowering of the gradient of the wave function (more precisely, its $z$ component in the bond axis direction) which entails a lowering of the corresponding component $T_{||}$ [42–44]. As a consequence, the $z$ component of the density gradient also decreases and $T_c$ has a similar behavior as $T^{\text{CI}}$: for all three systems the $T_c$ value at intermediate $R$ is lower than at larger and equilibrium $R$. For $T_{\text{HFF}}$ Eq. (57) holds true if $E_0^N$ is replaced with the Hartree-Fock total energy $E_{\text{HFF}}$. However, in this case a negative contribution of the second term is overcompensated with the gradual decrease of $E_{\text{HFF}}$ at larger distances, so that for the distances considered $T_{\text{HFF}}$ monotonically decreases with the increasing $R$.

The comparative features of $T_c$, $T^{\text{HF}}$, and $T^{\text{CI}}$ discussed above determine those of their differences $T_c$ and $T_c^{\text{CI}}$. Because of Eq. (56), both $T_c$ and $T_c^{\text{CI}}$ are always positive and because of the left-side inequality of (56) $T_c$ is consistently lower than $T_c^{\text{CI}}$. The molecular dissociation has a strikingly different effect on $T_c^{\text{CI}}$ and $T_c$. Due to the near-degeneracy effect (which is not taken into account in the restricted Hartree-Fock method), the left-right correlation is strengthened at larger bond distances, which causes an increase of $T_c^{\text{CI}}$. Contrary to this, in the dissociation limit $T_c$ approaches to the sum of the $T_c$ contributions of the atomic fragments, which is lower than the molecular $T_c$ value at $R_c$.

In particular, for the one-electron H atom $T_c^{\text{CI}} = 0$, since in this case the KS system coincides with the exact one, so that for the H₂ molecule $T_c$ should approach zero in the dissociation limit and for XH $T_c$ should approach the corresponding value for the individual atom X. This zero asymptotics for H₂ can be easily derived, if we employ the fact that in this limit the H₂ molecule is described properly with the Heitler-London wave function $\Psi_{\text{H}}(\vec{x}_1,\vec{x}_2)$

$$\Psi_{\text{H}}(\vec{x}_1,\vec{x}_2) = \frac{a^*(\vec{r}_1)b(\vec{r}_2) + a^*(\vec{r}_2)b(\vec{r}_1)}{2(1+S_{ab})^{1/2}} \times[\alpha(s_1)\beta(s_2) - \alpha(s_2)\beta(s_1)], \quad (58)$$

where $a(\vec{r})$ and $b(\vec{r})$ are the 1s-type atomic orbitals located on the H atoms A and B, $\alpha(i)$ and $\beta(i)$ are the one-electron spin functions and $S_{ab}$ is the overlap integral

$$S_{ab} = \int a^*(\vec{r})b(\vec{r})d\vec{r}. \quad (59)$$

A convenient feature of the Kohn-Sham theory is that even in the dissociation limit the Kohn-Sham system of H₂ is properly described with the one-determinantal wave function $\Psi_s(\vec{x}_1,\vec{x}_2)$ formed from the 1s orbitals $a(\vec{r})$ and $b(\vec{r})$

$$\Psi_s(\vec{x}_1,\vec{x}_2) = \frac{[a^*(\vec{r}_1) + b^*(\vec{r}_1)](a(\vec{r}_2) + b(\vec{r}_2))}{2\sqrt{2}(1+S_{ab})} \times[\alpha(s_1)\beta(s_2) - \alpha(s_2)\beta(s_1)], \quad (60)$$

since $\Psi_s(\vec{x}_1,\vec{x}_2)$ still generates the ground-state density $\rho(\vec{r})$ in this limit. Subtracting the kinetic energy obtained with $\Psi_s(\vec{x}_1,\vec{x}_2)$ from that obtained with $\Psi_{\text{H}}(\vec{x}_1,\vec{x}_2)$, one can derive an asymptotic expression for $T_c$

$$T_c = \langle \Psi_{\text{H}}|\hat{T}|\Psi_{\text{H}}\rangle - \langle \Psi_s|\hat{T}|\Psi_s\rangle = \frac{2(T_{\text{a}}S_{ab} - T_{ab})(1-S_{ab})}{(1+S_{ab})(1+S_{ab}^2)}, \quad (61)$$

where $T_a$ and $T_{ab}$ are the kinetic integrals

$$T_a = -\frac{1}{2} \int a^*(\vec{r})\nabla^2 a(\vec{r})d\vec{r}, \quad (62)$$

$$T_{ab} = -\frac{1}{2} \int a^*(\vec{r})\nabla^2 b(\vec{r})d\vec{r}. \quad (63)$$

The difference between total energies obtained with $\Psi_{\text{H}}(\vec{x}_1,\vec{x}_2)$ and $\Psi_s(\vec{x}_1,\vec{x}_2)$, the correlation energy $E_c$, approaches with increasing H—H distance its exact finite limiting value $-0.3125$ a.u. [31]. On the other hand $T_c$ the difference (61) between the corresponding kinetic energies decreases with the decreasing $S_{ab}$ and $T_{ab}$ at longer bond distances and it becomes zero in the infinite separation limit

$$T_c = 0, \quad R(\text{H—H}) \to \infty. \quad (64)$$

The results of the calculations presented in Tables I–III agree with the asymptotics derived theoretically. For H₂ the $T_c$ value calculated at $R(\text{H—H}) = 5.0$ a.u. is nearly twice as small as the corresponding value at $R(\text{H—H}) = 3.0$ a.u. (See Table III). In a similar way, for XH the $T_c$ values calculated at the largest distance considered are the least ones (See Tables I and II) and they are rather close to the $T_c$ values 0.038 a.u. and 0.095 a.u. obtained for the atoms Li and B, respectively, in [12]. As a consequence, in all cases the ratio $T_c/T_{c,H\text{FF}}$ gradually decreases with increasing bond distance. The local behavior of the potential $u_{\text{xc,kin}}$ which determines $T_c$ via the integral (54), as well as the form of other potentials will be discussed in the next section.

VI. $u_{\text{xc}}$ AND ITS COMPONENTS

Figures 2–4 compare the molecular Kohn-Sham exchange-correlation potentials $u_{\text{xc}}$ and the potentials of the exchange-correlation hole $u_{\text{hole}}^{\text{xc}}$ constructed for H₂ and XH.
understandable, since \( v_{xc}^{\text{hole}} \) represents the main correlation effect of reduction of the electron-electron repulsion due to the formation of the exchange-correlation hole. According to Eq. (35), \( v_{xc} \) is formed by the addition of \( v_{c, \text{kin}} \) and \( v_{\text{resp}} \) to \( v_{xc}^{\text{hole}} \), and the former potentials represent the repulsive effect of the ‘perturbation’ of the \((N-1)\)-electron system by the reference electron and the repulsive kinetic effect, respectively.

As expected, the molecular dissociation has relatively little effect on the form of \( v_{xc} \) and \( v_{xc}^{\text{hole}} \) in the inner region of atom X (See Figs. 3, 4). Both \( v_{xc} \) and \( v_{xc}^{\text{hole}} \) in this region have a deep well around the nucleus X, which represents the self-interaction part of the exchange potential of the 1s electron. At positions \( r \) within the 1s shell the exchange hole surrounding \( r \) is very close to minus the 1s density. Between the core and valence regions \( v_{xc} \) exhibits characteristic local maxima (intershell peaks). For BH one can see these peaks on both inner (with respect to the bond) and outer sides of the B atom and dissociation makes the outer-side peak more pronounced (See Fig. 4).

\( v_{xc} \) for LiH exhibits a similar peak only on the inner side of the Li atom (See Fig. 3), while on the outer side the intershell region is characterized only by a change of the slope of \( v_{xc} \). Dissociation makes this “peak” smaller, so that at large distance \( R = 7.0 \) a.u. \( v_{xc} \) in this region looks more like \( v_{xc} \) of the individual Li atom, which virtually lacks the intershell peak (See Fig. 2 of Ref. [12]). Unlike \( v_{xc}^{\text{hole}} \), \( v_{xc} \) is a more smooth potential: for both XH systems \( v_{xc}^{\text{hole}} \) is a monotonical function of \( z \) in the intershell regions for larger \( R \), while for \( R_e \) it exhibits much more shallow local maxima, which are displaced towards the bond midpoint.

The molecular dissociation manifests itself in the formation of a characteristic peak of \( v_{xc} \) near (or at) the bond midpoint \( z = 0 \). The H\(_2\) molecule provides an extreme example of such a peak (See Fig. 2). In this case \( v_{xc} \) already has a small bond midpoint peak for \( R_e = 1.401 \) a.u. However, it increases dramatically with increasing bond length, the corresponding maximum of \( v_{xc} \) is close to zero. The peak grows both in absolute value and with respect to \( v_{xc}^{\text{hole}} \), so that one can consider a formation of the bond midpoint peak on top of \( v_{xc}^{\text{hole}} \).

In the case of XH dissociation also creates, though visually less spectacular, a rather shallow peak of \( v_{xc} \) in the bonding region. It is displaced from \( z = 0 \) towards the H nucleus: for LiH at \( R = 7.0 \) a.u. \( v_{xc} \) reaches a local maximum at \( z = 0.70 \) a.u., while for BH at \( R = 5.0 \) a.u. this maximum is at \( z = 0.59 \) a.u. Unlike for the H\(_2\) molecule, for both XH molecules the position of the peak does not coincide with the corresponding local minimum of the density \( \rho \) for the interatomic part of the bond axis, the latter being placed at \( z = -0.05 \) a.u. for LiH and at \( z = 0.41 \) a.u. for BH.

Formation of the peak is accompanied with formation of a local well of \( v_{xc} \) just beyond the inner-side intershell peak. The corresponding minimum of \( v_{xc} \) is placed at \( z = -0.86 \) a.u. for LiH at \( R = 7.0 \) a.u. and at \( z = -1.16 \) a.u. for BH at \( R = 5.0 \) a.u. and \( v_{xc}^{\text{hole}} \) nearly touches \( v_{xc} \) at these points. One can see from Figs. 3(b) and 4(b) that the above-mentioned structure of \( v_{xc} \) in the bonding region is built up on top of \( v_{xc}^{\text{hole}} \), the latter potential being a rather smooth function in this region. From this it follows that other parts of \( v_{xc} \), namely, \( v_{c, \text{kin}} \) and \( v_{\text{resp}} \) are responsible for this structure and we shall proceed with the analysis of these parts.

(X = Li, B) for the equilibrium \( R_e \) and larger bond distances. The potentials are plotted along the bond axis as functions of the distance \( z \) from the bond midpoint. In all cases both \( v_{xc} \) and \( v_{xc}^{\text{hole}} \) are negative functions, with \( v_{xc} \) being consistently less attractive than the corresponding \( v_{xc}^{\text{hole}} \). This is quite un-

FIG. 2. Kohn-Sham exchange-correlation potential and the potential of the exchange-correlation hole for H\(_2\). (a) \( R_e = 1.401 \) a.u. and (b) \( R = 5.0 \) a.u.

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Figure 5 compares the potentials $v_{c, \text{kin}}$ obtained for $\text{H}_2$, LiH, and BH at the same elongated bond distance $R = 5.0$ a.u. [note that for the two-electron system $\text{H}_2$ $v_{c, \text{kin}}$ reduces to the potential $v_{\text{kin}}$ of Eq. (22)]. In all cases $v_{c, \text{kin}}$ exhibits a positive peak in the bonding region, though for $\text{XH}$ these peaks are much smaller than that for $\text{H}_2$. The peak grows higher with increasing bond length. One can see this from Fig. 6 where $v_{c, \text{kin}}$ constructed at two different bond distances $R$ are compared in the region of the bond peak for $\text{H}_2$ and LiH.
The behavior of $v_{c,\text{kin}}$ in the bonding region reflects the effect of the left-right correlation of electrons of a single bond $X-H$ on the conditional amplitude $\Phi(s_1, x_2, \ldots, x_N|\vec{r}_1)$. According to its definition (28), the relatively small $v_{c,\text{kin}}(\vec{r}_1)$ is a difference of two bigger potentials $v_{\text{kin}}(\vec{r}_1)$ and $v_{s,\text{kin}}(\vec{r}_1)$. In its turn, the latter difference is determined according to Eqs. (22), (25) by the integrated difference of the conditional amplitude gradients $|\nabla_1 \Phi(s_1, x_2, \ldots, x_N|\vec{r}_1)|^2$ and $|\nabla_1 \Phi_s(s_1, x_2, \ldots, x_N|\vec{r}_1)|^2$ or, in other words, by the relative sensitivity of the exchange-correlation and exchange (Fermi) holes in the distribution of other electrons to the displacement of the reference electron from $\vec{r}_1$. If $\vec{r}_1$ is in the bonding region and the reference electron is displaced from $\vec{r}_1$ towards a certain atom, another electron of the single bond gets an increase of its probability distribution on the other atom due to the left-right Coulomb correlation. This causes a change of the exchange-correlation hole associated with $\Phi(s_1, x_2, \ldots, x_N|\vec{r}_1)$ and produces a certain positive value of the amplitude gradient $|\nabla_1 \Phi(s_1, x_2, \ldots, x_N|\vec{r}_1)|^2$. Since there is no analogous exchange effect, the resulting $v_{c,\text{kin}}$ is definitely positive in this region. For the homatomic $H_2$ molecule $v_{c,\text{kin}}$ attains a maximum just at the bond midpoint $z=0$, while for $XH$ the corresponding maxima are displaced towards the H atom.

The left-right correlation is strengthened at larger bond distances by the strong near-degeneracy effect. As a consequence, the bond peak of $v_{c,\text{kin}}$ grows higher and Fig. 6 clearly illustrates this trend. In the case of $H_2$ $v_{c,\text{kin}}$ provides a dominating contribution to the bond peak of $v_{xc}$ at large distances $R$. For $XH$ $v_{c,\text{kin}}$ also makes a substantial contribution to the bond peak of $v_{xc}$, though in this case the corresponding contribution of the response potential $v_{\text{resp}}$ is some-
for $R_c$ (See Table III). An evident reason for this is that the peak arises in the bond midpoint region, which is the region of a low density $\rho$ at large $R(\text{H—H})$. The same trend, though not so spectacularly expressed, holds true also for the monohydrides XH.

While for all systems $v_{c,\text{kin}}$ vanishes in the region of the H atoms, it displays an oscillating behavior in the region of the X atoms of XH (See Fig. 5). The oscillations tend to be more contracted for the heavier atom B, the most visible feature being the positive peaks in the intershell regions. This oscillating behavior reflects different relative sensitivity of the exchange-correlation and Fermi holes associated with being the positive peaks in the intershell regions. This oscillation in a self-consistent way as the difference between the potential $\Phi_i(\vec{r}_1, \vec{r}_2, \ldots, \vec{r}_N)$ and $\Phi_i(\vec{r}_1, \vec{r}_2, \ldots, \vec{r}_N)[\vec{r}_1]$ to the displacement of the reference electron from $\vec{r}_1$ for different positions $\vec{r}_1$ in this region. The interpretation of this complicated behavior will be given elsewhere. We have no explanation for the sharp dip at the nuclei, which is possibly caused by the unphysical Gaussian shape of our CI density at the nuclei.

In Figs. 7, 8 the response potentials $v_{\text{resp}}$ are plotted, which have been obtained by subtraction of $v_{xc}^{\text{hole}}$ and $v_{c,\text{kin}}$ from $v_{xc}$. To make the interpretation of its form more clear, in Fig. 7 $v_{\text{resp}}$ is compared with the model potential $v_{\text{resp}}^{\text{mod}}$ [45]

$$v_{\text{resp}}^{\text{mod}}(\vec{r}) = \sum_{i=1}^{N} w_i \frac{|\phi_i(\vec{r})|^2}{\rho(\vec{r})},$$

(65)

the latter being the statistical average of the orbital contributions $w_i$

$$w_i = K \sqrt{\mu - e_i}.$$

(66)

The potential (65,66) with $K=0.38$ models the response part of the Krieger-Li-Iafrate (KLI) [46] approximation $v_{xc}^{\text{KLI}}$ to the exchange potential $v_{xc}^{\text{OPM}}$ of the optimized potential model (OPM) [47–49]

$$v_{xc}^{\text{KLI}}(\vec{r}) = v_{xc}^{\text{hole}}(\vec{r}) + \sum_{i=1}^{N} w_i \frac{|\phi_i(\vec{r})|^2}{\rho(\vec{r})}.$$

(67)

The parameters $w_i$ were defined within the KLI approximation in a self-consistent way as the difference between the expectation values of the potential (67) and the Hartree-Fock exchange operator $v_{xi}$ for the orbital $\phi_i$

$$w_i = \int \left|\phi_i(\vec{r})\right|^2 [v_{xc}^{\text{KLI}}(\vec{r}) - v_{xi}(\vec{r})] d\vec{r},$$

(68)

$$v_{xi}(\vec{r}_1) = -\frac{1}{\phi_i(\vec{r}_1)} \sum_{j=1}^{N} \phi_j(\vec{r}_1) \int \frac{\phi_i(\vec{r}_2) \phi_j(\vec{r}_2)}{|\vec{r}_1 - \vec{r}_2|} d\vec{r}_2,$$

(69)

and $v_{xc}^{\text{hole}}(\vec{r})$ in (67) is the potential of the Fermi hole, the exchange-only analogue of $v_{xc}^{\text{hole}}(\vec{r})$

$$v_{xc}^{\text{hole}}(\vec{r}) = \sum_{i=1}^{N} v_{xi}(\vec{r}) \frac{|\phi_i(\vec{r})|^2}{\rho(\vec{r})}.$$

(70)

The model (65,66) has the same orbital structure as the KLI construction and it satisfies the same condition [46] of zero contribution of the HOMO to the numerator of (65)

$$w_N = 0.$$

(71)

As was shown in [45], for atomic systems $v_{\text{resp}}^{\text{mod}}$ reproduces the characteristic stepped form of the response part of $v_{xc}^{\text{OPM}}$ and its KLI approximation.

Due to its construction (65), $v_{\text{resp}}^{\text{mod}}$ has a clear stepped form that helps to visualize the regions of various MO’s (See Fig. 7). The region of the core 1s electrons of atom X is characterized by a high plateau of $v_{\text{resp}}^{\text{mod}}$, while beyond this region $v_{\text{resp}}^{\text{mod}}$ has a steep descent to low values and it vanishes in the region of the HOMO around the H atom. This short-range behavior of $v_{\text{resp}}^{\text{mod}}$ follows from the KLI condition (71), so that all KLI-like potentials of the form (65) with (71) are expected to vanish in a similar way in the region of the HOMO. $v_{\text{resp}}^{\text{mod}}$ displays the above-mentioned features at both equilibrium and larger distances [Compare Figs. 7(a) and 7(c)]. When comparing the form of the potential for LiH and BH, one can note in the latter case an additional “shoulder” of $v_{\text{resp}}^{\text{mod}}$ in between the core and the HOMO regions. This can be attributed to the occupied nonbonding MO of BH formed, mainly, from the 2s orbital of the B atom.

In the region of atom X $v_{\text{resp}}^{\text{mod}}$ agrees qualitatively with the constructed $v_{\text{resp}}$. Though far from perfect, the one-step structure can be recognized for $v_{\text{resp}}$ with higher values for the core electrons and lower values for other electrons. $v_{\text{resp}}$ and $v_{\text{resp}}^{\text{mod}}$ display a steep descent to low values in the same regions, which is especially true for larger bond distances [See Figs. 7(a) and 7(b)]. The average height of $v_{\text{resp}}^{\text{mod}}$ in the core region appears to be somewhat lower than the step height of $v_{\text{resp}}^{\text{mod}}$ calculated with $K=0.38$, the corresponding relative difference is larger for LiH.

The step pattern of $v_{\text{resp}}$ in the region of atom X is disturbed, mainly, by the cusps and wiggles near the nuclei. One of the possible reasons for these features can be the inclusion of the correlation effects. Other reasons can be the deficiency of the Gaussian basis set representation of the CI density at the nucleus and the performance of the numerical procedure of $v_{xc}$ construction. Because of the singularity of the derivative of $\rho$ at nuclei, it appears to be somewhat difficult, in general, to achieve high accuracy close to nuclei, which of course is virtually impossible with Gaussian orbital based densities [4,11,13,14].

Unlike in the core regions, $v_{\text{resp}}$ and $v_{\text{resp}}^{\text{mod}}$ have a very different behavior in the HOMO region. As was discussed above, $v_{\text{resp}}^{\text{mod}}$ vanishes in this region. Contrary to this, $v_{\text{resp}}$ passes through a local minimum and then develops a positive buildup around the H atom, which is a more electronegative atom for both LiH and BH. One can draw the conclusion that the constructed $v_{\text{resp}}$ possesses a true feature, which has been established for $v_{\text{resp}}$ of the heteroatomic molecule in Sec. III, namely, the positive buildup of $v_{\text{resp}}$ around the more electronegative atom. Though the theoretical results of Sec. III have been obtained for the asymptotical case of large distances $R(A—B)$, the same qualitative picture holds true for both larger and equilibrium distances [compare Figs. 7(a) and 7(c)].

Based on the present analysis, one can expect that the KLI potential (67) lacks the positive buildup around the more electronegative atom. Indeed, the $v_{xc}^{\text{hole}}$ term of (67) has a similar smooth form as $v_{xc}^{\text{hole}}$ presented here, while the sec-
second term of (67) has the same structure as $v_{\text{resp}}^\text{mod}$ and the latter potential demonstrates the absence of the above-mentioned feature. When comparing $v_{xc}$ and $v_{xc}^{\text{KLI}}$, the positive build-up appears as a correlation effect for the heteroatomic molecules, which is present in the exchange-correlation potential $v_{xc}$ and is absent in the exchange-only potential $v_{xc}^{\text{KLI}}$. Also the bond midpoint peak due to $v_{c,\text{kin}}$ would be absent in $v_{xc}^{\text{KLI}}$.

The inclusion of the positive buildup is essential in order to provide the proper energies $\epsilon_N$ of the HOMO. Because of its absence, the HOMO energies obtained for LiH and BH with the combined potential

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**FIG. 7.** Comparison of the potentials $v_{\text{resp}}$ and $v_{\text{resp}}^\text{mod}$ (a) LiH, $R = 7.0$ a.u. and (b) BH, $R = 5.0$ a.u. and (c) LiH, $R_e = 3.015$ a.u.
appears to be always at more negative energies than the corresponding \( v_N \) values obtained with the constructed \( v_{xc} \) (the latter being presented in Tables I and II). Variation of the parameter \( K \) within reasonable limits failed to improve this deficiency.

Figure 8(a) compares the potentials \( v_{resp} \) obtained for \( H_2 \), LiH, and BH at the same elongated bond length \( R = 5.0 \) a.u. It can serve as an illustration to the formulas (43)–(46) derived for \( v_{resp} \) in Sec. III. For both LiH and BH \( v_{resp}(\vec{r}) \) in the region around the H atom displays the positive buildup which, according to Eqs. (43) and (45), is represented by the first three terms of Eq. (43). When the position \( \vec{r} \) of the reference electron moves to the region of the HOMO near the bond midpoint \( z=0 \), the above-mentioned terms vanish and \( v_{resp} \) displays a minimum, which is represented by Eq. (44). When the reference electron moves further into the inner region of atom X, one has subtract from the single term of Eq. (44) the analogous contribution from \( v_N^{(N-1)} \) of Eq. (34). However, the energetical effect \( \Delta E(A^+ - B; \vec{r}) \) of the electron redistribution in \( A^+B \) due to the presence of the reference electron in this inner region is much larger than that for the HOMO region and, as a result, the stepped structure of \( v_{resp} \) is created in the former region. Contrary to this, in the case of the two-electron homoatomic \( H_2 \) molecule \( v_{resp} \) is rather small everywhere and it has a flat form with very shallow maxima at the H nuclei. This flat form is anticipated, since in all regions the same formula (46) is valid for \( H_2 \).

In Fig. 8(b) \( v_{resp} \) obtained for LiH at \( R = 5.0 \) a.u. and \( R = 7.0 \) a.u. is plotted in the region of the H atom, the latter being placed at \( z = 0 \). One can see from Fig. 8(b) that the positive buildup around the H atom has a similar form for both distances. In accordance with the one-dimensional model and theoretical considerations of Sec. III, the positive buildup grows higher with the increasing \( R \) and its maximum comes closer to the difference \( \Delta I_p = 0.302 \) a.u. between the ionization energies of the H and Li atoms.

**VII. CONCLUSIONS**

In this paper the effect of molecular dissociation on the exchange-correlation Kohn-Sham potential \( v_{xc} \) has been established and analyzed. \( v_{xc} \) and its components have been constructed from *ab initio* correlated first- and second-order density matrices for the heteroatomic molecules LiH and BH at several bond distances \( R(X-H) \). The results have been compared with those for the two-electron homoatomic molecule \( H_2 \).

The molecular dissociation manifests itself in the formation of a characteristic peak of \( v_{xc} \) in the bonding region. In order to interpret this behavior, a partitioning of \( v_{xc} \) has been used employing the partially integrated conditional probability amplitude \( \Phi(s_1, x_2, \ldots, x_N | r_1) \). This partitioning represents \( v_{xc} \) as a sum of the potential of the exchange-correlation hole \( v_{xc}^{\text{hole}} \), the kinetic component \( v_{c,kin} \) and the “response” potential \( v_{resp} \). For the homoatomic \( H_2 \) molecule the peak of \( v_{xc} \) is determined by the bond midpoint peak of \( v_{c,kin} \) (which in this case reduces to \( v_{kin} \)), while for the heteroatomic \( XH \) molecule the peak of \( v_{xc} \) is a combination of the bonding peak of \( v_{c,kin} \) and the positive buildup of \( v_{resp} \) around the more electronegative atom H. In all cases the peak of \( v_{xc} \) and the corresponding features of \( v_{c,kin} \) and \( v_{resp} \) grow higher with increasing bond distance.
It has been established, using the representation of $v_{\text{rep}}$ in terms of the effective potential $v^{N-1}$ of the $(N-1)$ electron system introduced in [2], that the positive buildup of $v_{\text{rep}}$ originates from the difference between the electron distribution of $(N-1)$ electrons associated with the conditional amplitude $\Phi(x_1,x_2,\ldots,x_N|x_1)$ of the heteroatomic molecule $AB$ and that corresponding to the ground-state wave function of the cation $(AB)^+$. From the conditional amplitude analysis the asymptotical expressions for $v_{\text{rep}}$ and its positive buildup have been obtained. The latter is represented as a leading term $\{T_{\text{xc}} - T_{\text{xc}}^0\}$, which arises due to the different ionization energies of atoms $A$ and $B$, plus corrections due to the different polarization of $A$ and $B$ by a positive charge and different “response” of electrons of the cations $A^--B^+$ and $A^+-B$ to the presence of the additional reference electron.

The dependence of the kinetic energy of noninteracting particles $T_e$, the kinetic part of the correlation energy $T_{\text{xc}}$, and the energy of the HOMO $\epsilon_N$ on the bond distance $R$ has been studied. In all cases $T_e$ as a function of $R$ passes through a minimum, while $T_{\text{xc}}$ and $\epsilon_N$ decreases for large $R$, both approaching the corresponding values for the individual atoms $X$. In the particular case of the $H_2$ molecule an accurate asymptotic formula for $T_e$ has been obtained, according to which it approaches zero in the bond dissociation limit.

An important problem that still remains is how to increase further the numerical accuracy and stability of the molecular $v_{\text{xc}}$ construction. As has been discussed above, construction in the finite Gaussian basis suffers from the eventual development of artificial oscillations of $v_{\text{xc}}$ and from an inadequacy of a Gaussian basis both at nuclei and at molecular density tails. These difficulties can be, at least in principle, overcome by the construction of $v_{\text{xc}}$ within a basis-set-free numerical molecular program. This problem, as well as the construction of $v_{\text{xc}}$ and the exchange-correlation density $\epsilon_{\text{xc}}$ for more complex molecular processes will be addressed in our further work.

[41] I. N. Levine, Quantum Chemistry (Allyn and Bacon, Boston, 1983).